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Assessment of physico-chemical properties of black cotton soils from different blocks of Guntur District, Andhra Pradesh, India

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Abstract

The present investigation was carried at Sam Higginbottom University of Agriculture Technology and Sciences to assess the Physicochemical properties of black cotton soils from different blocks of Guntur district, Andhra Pradesh, India. A total of twenty-seven soil samples were collected randomly from different depths, i.e., 0-15cm,15-30 cm, and 30-45cm. The study area consists of mostly black cotton soil. These soils were moderate to strongly alkaline in reaction and non-saline. On the soil complex, the dominant cation is calcium. The overall fertility status of the soils was low, medium, and high in nitrogen, phosphorus, and potassium respectively. The calcium and magnesium ranges are high in these clay soils. The sulfur is sufficient in these clay soils. As the soils were calcareous and strongly alkaline, there is a need for the application of any acid-forming amendment (S containing amendments) and organic materials to alleviate the nutrient deficiency and improve productivity.

Keywords: Physico-chemical properties, Alkaline, Water retaining capacity, Black cotton soils

1. Introduction

Soil is the backbone of our food security. Without healthy soils, farmers wouldn't be able to provide us with feed, fiber, food, and fuel. Our farmers need to understand the components which make up the soil in which their crops grow. Adequate crop growth and its production mainly depend on the appropriate nutrition, if there is a nutrient deficiency in the soil it affects the growth rate of plants.

Nitrogen occupies the first position in the plant requirement among the nutrient elements, followed by phosphorus and potassium (Samuel and Ebenezer, 2014; Solanki and Chavda, 2012) ^[19, 24]; Potassium is a major nutrient that plays a major role in different physiological processes of plants helping plants to resist against diseases and improving physical characteristics of the plant. Magnesium is necessary for the synthesis of chlorophyll pigment in green plants and its deficiency causes the loss of healthy green color of leaves (Mahajan and Billore, 2014) ^[13, 20]. Calcium ion is the key element in reducing the soil salinity erosion content and as well as phosphorous loss through flowage. Phosphorus is the most important element because the growth of plants depends on the availability of Phosphorous content in the soils. Soil fertility and nutrient management are important factors that have a direct impact on crop yield and quality.

To identify the fertility status of the selected area, various soil samples were collected from pre-determined locations and were analyzed for Physico-chemical properties (pH and electrical conductivity) chemical characteristics including fertility parameters like available nitrogen, phosphorous, potassium, sulfur, and exchangeable basic cations constituting calcium, magnesium.

2. Materials and Methods

2.1 Study area

The location of the Guntur district lies between 16^{0} 30'67" N latitude and 80^{0} 43'65" E longitude. It covers a geographical area of 11,391 sq km. (Fig.1). The Krishna River forms the northeastern and eastern boundary of the district, separating Guntur District from Krishna District. It is located near the Bay of Bengal and is surrounded by many suburban areas. Guntur district experiences a tropical climate in summer. And the dry and cold climate in winter. The maximum temperature is 32 °C and the minimum temperature is 20 °C. The average annual temperature is 28.5 °C.

When compared with winter, the summers have much more rainfall. In a year, the average rainfall recorded is 906 mm.

2.2 Sample collection

Soil samples were collected from three different blocks of the Guntur district of Andhra Pradesh. They are Pedakurapadu

Krosuru, Sattenpalli. Soil samples were collected with the help of Khurpi, spade, and meter scale. In each block, three villages were selected for sampling and were samples collected randomly from different depths i.e., 0-15cm,15-30cm, and 30-45cm. A total of twenty-seven soil samples were collected.

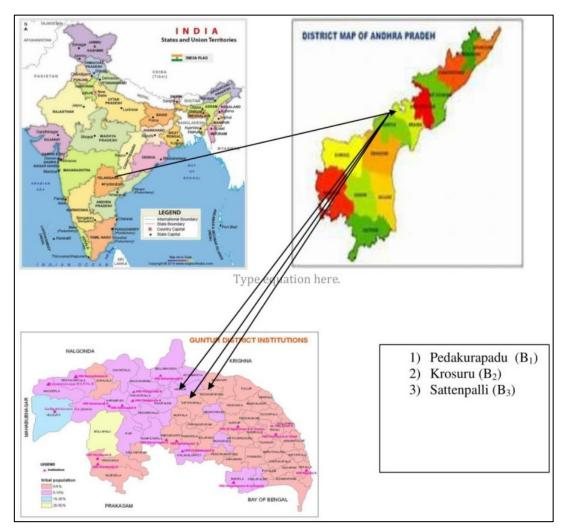


Fig 1: Map of the study area

2.3 Soil analysis

The pH was determined in 1:2 soil water suspensions using digital pH meter (Jackson, 1958). The EC was determined in 1:2 soil water suspensions using digital EC meter (Wilcox, 1950) [30]. The soil was distilled with alkaline potassium permanganate as suggested by (Subbiah and Asija 1956) [25] and the ammonia evolved was determined. P in the soil extract is determined colorimetrically using a Photoelectric Colorimeter after developing molybdenum blue colour (Olsen et al., 1954)^[18]. The procedure was based on extraction with 1N NH4OAC (pH 7.0) and K was determined by Flame Photometer (Toth and Prince, 1949) ^[27]. The same procedure used for the estimation of K. Exchangeable calcium and magnesium was determined by 1N Neutral Ammonium Acetate Saturation Method or EDTA method as laid out by Cheng and Bray (1951). Available sulphur was estimated by the turbidimetric method as put forth by Bardsley and Lancaster (1960) [2].

3. Results and Discussion

3.1 pH and EC (ds m⁻¹)

Table 1. depicted the statistical accumulation on pH and EC

of various farmer's fields which was found to be significant differences due to depth and site. The pH ranges from 8.01 to 8.79. The highest mean value is recorded 8.79 in B3V2 and the least mean value 7.50 in B2V3. The EC ranges from 0.37 to 0.81 ds m⁻¹. The highest mean value is recorded at 0.81 ds m⁻¹ in B2V1 and the least mean value of 0.37 ds m⁻¹ in B3V2. Higher values were recorded in deeper layers. A similar trend was observed by Dhale and Jagdish Prasad, (2009) ^[9] in the black soil of Jalna district, Maharashtra.

3.2 Organic carbon and organic matter

Table 2. depicted the statistical accumulation on Organic Carbon and Organic matter of various farmer's fields which was found to be significant differences due to depth and site. The Organic carbon ranges from 0.35 to 0.70%. The highest mean value is recorded 0.70% in B2V1 and the least mean value 0.35% in B3V2. The OM ranges from 0.60 to 1.16%. The highest mean value 0.67% in B2V2. When depth-wise values were considered, lower values were recorded in deeper layers. Nayak *et al.* (2002) observed a similar organic carbon range (0.11 to 0.82 percent) in black soils of the Indo- Gangetic

plains of West Bengal.

3.3 Available Nitrogen, Phosphorus, and Potassium

Table 3. depicted the statistical accumulation of Nitrogen (kg ha⁻¹) and Potassium (kg ha⁻¹) of various farmer's fields and depths which were found to be significant differences due to depth and site but phosphorus showed nonsignificance due to depth and significance due to site. The N ranges from 168 to 277.66(kg ha⁻¹). The highest mean value is recorded 277.66 in B2V1 and the least mean value 168. Similar trends were observed by Bharmbe et al. (1999)^[4] in Vertisols of the Majalgao canal command area. The Phosphorus ranges from 11 to 60.3(kg ha⁻¹) 3. The highest mean value is recorded 60.33(kg ha⁻¹) in B1V1 and the least mean value 11(kg ha⁻¹) in B3V1. Satish (2003)^[21] and Varaprasad Rao et al. (2008) ^[28] reported medium availability of phosphorus in soils of Chebrolu Mandal, Guntur district, and Ramachandrapuram Mandal, Chittoor districts of Andhra Pradesh, respectively. The Potassium ranges from 505 to 984.33(kg ha⁻¹). The highest mean value is recorded at 984.33 in B1V1 and the least mean value 479(kg ha⁻¹) in B2V3. Similar observations of high potassium content were reported by Bandyopadhyay et al. (2004) ^[1] and Dhale and Jagdishprasad (2009) ^[9] in black soils of Maharashtra. The available Nitrogen, Phosphorus, and Potassium content were high values at the surface than in lower layers.

3.4 Exchangeable Calcium and Magnesium

Table 4. depicted the statistical accumulation of exchangeable calcium [cmol (p+) kg⁻¹] showed no significant difference was found at depth and a significant difference was found at villages. Very low values were recorded in all the sites. This may be due to the leaching of calcium as hydrogen is added to the soil by the decomposition of organic matter as well as due to heavy rainfall. The highest mean value of 27.7 cmol (p+) kg⁻¹ was recorded at B3V3. And the lowest mean value of 22.76 cmol (p+) kg⁻¹ was recorded at B3V2. Magnesium showed No significant difference in both depth and site. Very low values were recorded in all the sites. This may be due to the leaching of magnesium as hydrogen is added to the soil by the decomposition of organic matter as well as due to the leaching of magnesium as hydrogen is added to the soil by the decomposition of organic matter as well as due to the leaching of magnesium as hydrogen is added to the soil by the decomposition of organic matter as well as due to the leaching of magnesium as hydrogen is added to the soil by the decomposition of organic matter as well as due to heavy rainfall. The highest mean value of 9.9 cmol (p+) kg-1 was

recorded in B1V1. The maximum exchangeable magnesium of 11.73 cmol (p+) kg-1 was recorded in B1V3 while the minimum value was recorded as 6.4 cmol (p+) kg-1 in B3V3. Similar results were observed by Naga Raju Kola and Babu Rao Gudipudi (2020) ^[16]. Soil Chemistry of Erravagu Subbasin of Guntur District, Andhra Pradesh 2020

3.5 Available Sulphur

Table 5. depicted the available sulfur (ppm) in soils from various villages and at different profile depths. A significant difference was found. Medium values of available sulfur were recorded in all the sites. The highest mean value was recorded at B2V1 as 36.66 ppm. Low values may be attributed to the leaching of sulfur. The available sulfur was found to decrease with an increase in depth. The maximum available sulfur was recorded in B2V1 which was 36.33 ppm. While the minimum value was recorded in B3V1 as 9.33 ppm. Similar trends were observed in Inceptisol of Chittoor district, Andhra Pradesh (Basavaraju *et al*, 2005) ^[3] and (Varaprasad Rao *et al*. 2008) ^[28].

3.6 Correlation Coefficient (R) Between Physicochemical Properties of Black Cotton Soils of Guntur District, Andhra Pradesh, India

The electrical conductivity showed the significant negative correlation with pH (-0.71 @ CD P = 0.01). The available nitrogen showed the significant positive correlation with EC (0.711 @ CD = 0.01). The available nitrogen showed nonsignificant and negative correlation with pH (-0.334). The available Phosphorus showed the significant positive correlation with% organic carbon (0.549 @ CD P = 0.05). The available Potassium showed the positive correlation with both% organic carbon $(0.576 \ @ \ CD \ P = 0.05)$ and Phosphorus (0.970 @ CD P = 0.01). The exchangeable calcium showed the significant positive correlation with available nitrogen (0.820 @ CD P = 0.01). The exchangeable magnesium showed the significant positive correlation with Phosphorus (0.764 @ CD P =0.01) and Potassium (0.716 @ CD P = 0.01). The available sulphur showed the positive correlation with both available Nitrogen (0.801@ CD P =0.01) and exchangeable Calcium (0.540 @ CD P = 0.05).

Table 1: Soil pH and Soil EC (dS m-1) at different depths (cm)

villagog		рН		EC				
villages	0- 15cm	15-30cm	30- 45cm	0-15cm	15-30cm	30- 45cm		
B1 V1	7.8	8.1	8.2	0.73	0.79	0.85		
B1 V2	8.36	8.43	8.53	0.39	0.43	0.47		
B1 V3	8.25	8.51	8.62	0.63	0.67	0.7		
B2 V1	8.3	8.42	8.51	0.77	0.82	0.86		
B2 V2	8.4	8.61	8.71	0.43	0.46	0.48		
B2 V3	7.9	7.95	8.2	0.63	0.69	0.75		
B3 V1	8.45	8.67	8.81	0.53	0.57	0.61		
B3 V2	8.62	8.82	8.91	0.33	0.37	0.42		
B3 V3	8.74	8.81	8.82	0.43	0.47	0.5		
Range	7.8-8.74	7.95-8.82	8.2-8.91	0.33- 0.77	0.37-0.79	0.43- 0.86		
Mean	8.31	8.48	8.59	0.54	0.58	0.62		
	F- test	S.Ed. (±)		F- test	S.Ed. (±)			
	S	0.080423		s	0.024704			
	S	0.094383		S	0.054379			

Table 2: Soil Organic carbon (%) and Soil Organic matter (%) at different depths (cm)

villages	O	ganic carbon (%)	Organic matter			
	0- 15cm	15-30cm	30- 45cm	0- 15cm	15- 30cm	30- 45cm	
B1 V1	0.72	0.68	0.63	1.24	1.17	1.08	

B1 V2	0.5	0.49	0.45	0.86	0.84	0.77
B1 V3	0.47	0.39	0.38	0.81	0.67	0.65
B2 V1	0.39	0.35	0.32	0.67	0.6	0.55
B2 V2	0.48	0.45	0.42	0.82	0.77	0.72
B2 V3	0.46	0.42	0.39	0.79	0.72	0.67
B3 V1	0.45	0.41	0.38	0.77	0.7	0.65
B3 V2	0.56	0.54	0.52	0.96	0.93	0.89
B3 V3	0.51	0.48	0.39	0.87	0.82	0.67
Range	0.39- 0.72	0.35-0.68	0.32-0.63	0.67-1.24	0.60-1.17	0.55-1.08
Mean	0.50	0.46	0.44	0.86	0.80	0.73
	F-test	S.Ed. (±)		F- test	S.Ed. (±)	
	S	0.02117		S	0.0365	
	S	0.031265		S	0.0539	

villages	Available Nitrogen (kg ha ⁻¹)			Available phosphorus (kg ha ⁻¹)			Available Potassium (kg ha ⁻¹)		
villages	0-15cm	15-30cm	30-45cm	0-15cm	15-30cm	30-45cm	0-15cm	15-30cm	30-45cm
B1 V1	238	225	217	63	60	58	996	987	970
B1 V2	201	189	175	53	51	67	932	927	918
B1 V3	220	212	207	48	47	45	900	893	887
B2 V1	289	275	269	17	17	15	604	597	593
B2 V2	238	233	225	19	18	16	585	578	565
B2 V3	207	201	193	18	17	14	503	497	479
B3 V1	204	193	182	13	11	9	513	507	495
B3 V2	180	169	157	17	15	13	617	659	647
B3 V3	209	197	183	23	21	19	653	647	635
Range	180-289	169-275	157-269	13 - 63	11-60	9-67	503-996	497-987	479-970
Mean	220.66	210.44	200.88	30.11	28.56	28.44	700.33	699.11	687.66
	F-test	S.Ed. (±)		F-test	S.Ed. (±)		F-test	S.Ed. (±)	C.D.at 0.05%
	S	5.710		NS	0.5379		s	4.0339	
	S	10.601		S	6.5735		S	62.599	

 Table 4: Exchangeable calcium and Magnesium [cmol (p+)kg-1]

	Exchangea	ble calcium [cm	ol (p+)kg-1	Exchangeable magnesium [cmol (p+)kg-1				
villages	0- 15cm	15- 30cm	30- 45cm	0- 15cm	15- 30cm	30-45cm		
B1 V1	22.5	27.5	26	10.5	9.9	9.3		
B1 V2	24.7	23.5	24.1	12.3	11.7	11.2		
B1 V3	27.8	24.3	26.1	11.3	10.5	10.3		
B2 V1	34.5	32.7	33.2	9.5	8.7	8.3		
B2 V2	26.3	24.3	23.2	10.2	9.8	9.3		
B2 V3	25.8	24	24.5	7.2	6.8	7.1		
B3 V1	24.9	23.5	23.2	7.9	7.8	7.7		
B3 V2	23.7	22.5	22.1	6.8	6.7	6.6		
B3 V3	28.9	27.4	26.8	6.5	6.3	6.4		
Range	23.7-34.5	22.5-32.7	22.1-33.2	6.5-12.3	6.3-11.7	6.4-11.2		
Mean	26.56	25.52	25.46	9.13	8.68	8.46		
	F-test	S.Ed. (±)		F-test	S.Ed. (±)			
	NS	0.3577		NS	0.1959			
	S	1.0605		NS	0.6306			

 Table 5: Available Sulphur (ppm)

	Available Sulfur (ppm)						
villages	0-15cm	15-30cm	30-45cm				
B1 V1	29	25	23				
B1 V2	26	24	21				
B1 V3	25	22	19				
B2 V1	39	37	33				
B2 V2	37	34	31				
B2 V3	30	27	23				
B3 V1	12	9	7				
B3 V2	18	15	12				
B3 V3	21	19	9				
Range	12-39	9-37	7-33				
Mean	26.33	23.55	19.77				
	F-test	S.Ed. (±)					
	S	1.8997					
	S	2.9239					

1 .721**	1							
	1							
0.050								
0.252 -0	0.061	1						
0.334 0.7	710**	-0.287	1					
0.431 0	.190	0.549*	-0.004	1				
0.268 0	.129	0.576*	-0.014	0.970**	1			
0.070 0.	592*	-0.456	0.820**	-0.131	-0.090	1		
0.297 0	.149	0.096	0.286	0.764**	0.716**	-0.007	1	
0.475 0	.446	-0.165	0.801**	0.112	0.056	0.540*	0.407	1
0.	.268 0 .070 0. .297 0	.268 0.129 .070 0.592* .297 0.149	.268 0.129 0.576* .070 0.592* -0.456 .297 0.149 0.096	.268 0.129 0.576* -0.014 .070 0.592* -0.456 0.820** .297 0.149 0.096 0.286	.268 0.129 0.576* -0.014 0.970** .070 0.592* -0.456 0.820** -0.131 .297 0.149 0.096 0.286 0.764**	.268 0.129 0.576* -0.014 0.970** 1 .070 0.592* -0.456 0.820** -0.131 -0.090 .297 0.149 0.096 0.286 0.764** 0.716**	.268 0.129 0.576* -0.014 0.970** 1 .070 0.592* -0.456 0.820** -0.131 -0.090 1 .297 0.149 0.096 0.286 0.764** 0.716** -0.007	.268 0.129 0.576* -0.014 0.970** 1 .070 0.592* -0.456 0.820** -0.131 -0.090 1 .297 0.149 0.096 0.286 0.764** 0.716** -0.007 1

* Significant at (0.05) 5% level; ** Significant at(0.01) 1% level.

EC= Electrical Conductivity, OC=Organic Carbon, N=Available Nitrogen, P=Available Phosphorus, K=Available Potassium, Ca=Exchangeable Calcium, M= Exchangeable Magnesium, S= Available Sulphur.

4. Conclusion

It is concluded that the study area consists of black cotton soil. These soils were moderate to strongly alkaline in reaction and non-saline. On the soil complex, the dominant cation is calcium. The overall fertility status of the soils was low, medium, and high in nitrogen, phosphorus, and potassium respectively. The calcium and magnesium ranges are high and sulfur is sufficient in these clay soils. These analyses may help the farmers to maintain proper nutrient management and as the soils were calcareous and strongly alkaline, there is a need for the application of any acidforming amendment (S containing amendments) and organic materials to alleviate the nutrient deficiency and improve productivity.

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6. References

- 1. Bandyopadhyay KK, Ghosh PK, Chaudhary RS, Mhati K, Mandal KG, Misra AK. Integrated nutrient management practices in soybean and sorghum in sole and intercropping system in a Vertisol. Indian Journal of Agricultural Science 2004;74:55-63.
- 2. BardsLey CE, Lan Caster ID. Determination of reserve sulfur and soluble sulfates in 1960.
- Basavaraju D, Naidu MVS, Ramavatharam N, Venkaiah K, Ramarao G, Reddy KS. Characterisation and evaluation of soils in Chandragiri Mandal of Chittoor district, Andhra Pradesh. Agropedology 2005;15:55-62.
- 4. Bharambe PR, Kadam SG, Shinde SG, Shelke DK. Characterization of soils of Majalgao canal Command area. Journal of the Indian Society of Soil science 1999;47:749-754.
- 5. Black CA. Methods of soil analysis. American Society of Agronomy, Madison, Wisconsin, USA 1965, 2.
- 6. Bouyoucos GJ. The hydrometer is a new method for the mechanical analysis of soils 1927.
- 7. Brady NC, Weil RR. The nature and properties of soils, Eleventh Edition, Prentice- Hall, New York 1996.
- 8. Das DK. *Introductory Soil Science*, 2nd Edition. Kalyani Publisher. New Delhi 2004
- 9. Dhale SA, Jagdish Prasad. Characterization and Classification of Sweet Orange- growing Soils of Jalna

District, Maharashtra. Journal of the Indian Society of Soil Science 2009;57(1):11-17.

- Fisher RA. Statistical methods and scientific induction. Journal of the royal statistical society series. 1925;17:69-78.
- Jackson ML. Soil Chemical Analysis. Open Journal of Soil Science 1973;5(4)
- 12. Kusumakumari T, Sreenevasulu A, Rao PS. Soil fertility survey of forest soils of Guntur district. Indian journal of the environmental protocol 2011;1(9):850-851.
- Mahajan S, Billore D. Assessment of physico-chemical characteristics of the soil of Nagchoon pond Khandwa, MP, India. Research J. of Chemical Sci 2014;4(1):26-30.
- Munsell AH. Munsell Soil Color Charts. Munsell Color Company Inc 1954.
- 15. Baltimore. Muthuvel P, Udayasoorian C, Natesan R, Ramaswami PR. Introduction to Soil Analysis, Tamil Nadu Agricultural University, Coimbatore 1992.
- 16. Naga Raju Kola, Babu Rao Gudipudi. Soil Chemistry of Erravagu Sub-basin of Guntur District, Andhra Pradesh 2020.
- Nayak AK, Chinchmalatpure RA, Gururaja Rao G, Verma AK. Swell shrink potential of Vertisols in relation to clay content and exchangeable sodium under different ionic environments. Journal of the Indian Society of Soil Science 2006;54:1-5.
- Olsen SR, Cole CV, Watanabe FS, Dean LA. Estimation of available phosphorus in soils by extraction with sodium bicarbonate. U.S. Dep. of Agric. Circ. 939.P.L.G. (Eds.), Land Useand Soil Resources 1954, 9-22.
- 19. Samuel AL, Ebenezer AO. Mineralization rates of soil forms of nitrogen, phosphorus, and potassium as affected by organomineral fertilizer in sandy loam. Advances in Agriculture, 2014, 5.
- Saroj Mahajan, Dilip Billore. Assessment of Physico-Chemical Characteristics of the Soil of Nagchoon Pond Khandwa, MP, India. Res. J Chem Sci 2014;4(1):26-30.
- Satish A. Studies on land conditions of vegetable growing belt in Narakoduru area of Andhra Pradesh. MSc. (Ag.) thesis submitted to Acharya N.G. Ranga Agricultural University, Rajendranagar, Hyderabad 2003
- 22. Singh S, Singh JS. Microbial biomass is associated with water-stable aggregates in the forest, savanna, and cropland soils of a seasonally dry tropical region, India. Soil Biology and Biochemistry 1995;27:1027-1033.
- Smith K. Soil organic carbon dynamics and land-use change. In: Braimoh, A.K., Vlek, Soil Science, 2008;23:343-353. soils. Soil Sci. Soc. Amer. Proc., 24: 255-268.
- 24. Solanki HA, Chavda NH. Physicochemical analysis with

reference to seasonal changes in soils of Victoria park reserve forest, Bhavnagar (Gujarat). Life sciences Leaflets 2012;8:62-68

- 25. Subbiah BV, Asija GL. A rapid procedure for the determination of availablenitrogen in the soil, Curr. Sci. 1956;25:259-260.
- 26. Thakre YG, Choudhary MD, Raut RD. Physico-chemical characterization of red and black soils of Wardha Region. International Journal of Chemical and Physical Sciences 2012;1(2):60-66.
- 27. Toth SJ, Prince AL. Estimation of cation exchange capacity and exchangeable Ca, K and Na content of soil by flame photometer technique. Soil Sci 1949;67:439-445.
- 28. Varaprasad Rao AP, Naidu MVS, Ramavatharam N, Rama Rao G. Characterization, classification and Evaluation of soils on different land forms in Ramachandrapuram mandal of Chittoor district in Andhra Pradesh for sustainable land use planning. Journal of the Indian Society of Soil Science 2008;56(1):23-33.
- 29. Walkley A, Black IA. Critical examination of a rapid method for determining organic carbon in soils, the effect of variance in digestion conditions and of inorganic soil constituents. Soil Sci 1947, 632-251.
- 30. Wilcox LV. Electrical conductivity. Am. waterworks Assoc. J 1950;42:775-776.