www.ThePharmaJournal.com

The Pharma Innovation



ISSN (E): 2277- 7695 ISSN (P): 2349-8242 NAAS Rating: 5.23 TPI 2021; SP-10(9): 407-410 © 2021 TPI www.thepharmajournal.com Received: 07-07-2021 Accepted: 09-08-2021

Trishanku Kashyap

Department of Soil Science and Agricultural Chemistry, Central Agricultural University, Iroisemba, Imphal, Manipur, India

Herojit Singh Athokpam

Department of Soil Science and Agricultural Chemistry, Central Agricultural University, Iroisemba, Imphal, Manipur, India

N Surbala Devi

Department of Soil Science and Agricultural Chemistry, Central Agricultural University, Iroisemba, Imphal, Manipur, India

Nakeertha Venu

Department of Soil Science and Agricultural Chemistry, Central Agricultural University, Iroisemba, Imphal, Manipur, India

K Nandini Devi

Department of Agronomy, Central Agricultural University, Iroisemba, Imphal, Manipur, India

N Gopimohan Singh

Department of Basic Science, Central Agricultural University, Iroisemba, Imphal, Manipur, India

Corresponding Author Trishanku Kashyap Department of Soil Science and

Agricultural Chemistry, Central Agricultural University, Iroisemba, Imphal, Manipur, India

Forms and status of phosphorus in acid soils of Kakching district, Manipur (India)

Trishanku Kashyap, Herojit Singh Athokpam, N Surbala Devi, Nakeertha Venu, K Nandini Devi and N Gopimohan Singh

Abstract

Twenty soil samples were collected from different locations of Kakching district, Manipur for this experiment. All the investigated soil samples were acidic in nature with a mean value pH 4.98. There was wide variation of organic carbon content with a mean value of 16.76 g kg⁻¹. The mean EC, Cation exchange capacity, available nitrogen, available potassium were 0.22 dSm⁻¹, 14.01 [cmol(P+)kg⁻ '], 212.66 kg ha⁻¹, 114.75 kg ha⁻¹, respectively. For the soil studied the amount of inorganic P fractions was in the order: Red-P> Fe-P> Al-P>Occl-P>Ca-P>Sal-P. Al-P and Fe-P were negatively correlated with pH (r = -0.446*) and (r=-0.447*) respectively. Saloid P was positively correlated with E.C. (r =0.512*) and CEC (r=0.456*) along with Red-P (r =0.555*). Occl-P was negatively correlated with available potassium (r = -0.505*) and positively and significantly correlated pH (r=0.583**)and CEC (r=0.577**). Ca-P was positively and significantly correlated with E.C (r = 0.500*), clay fraction (r = 0.583**) and CEC (r = 0.605**). Sal-P was positively correlated with Ca-P and negatively correlated with Org-P (r=0.520* and -0.547*). Fe-P was positively correlated with Tot-P (r=0.540*) and Red-P with Ca-P (r=0.628**). Finally, the Tot-P was positively correlated with Org-P (r=0.882**).

Keywords: P fractions, soil properties, correlation, acid soil

1. Introduction

Phosphorus is referred to as the "king-pin" in Indian agriculture and also as "energy currency" of the plants. It is one of the three primary nutrients and the second most important element after nitrogen for growth and development of every plants. Phosphorus aids in plant growth by enabling photosynthesis (energy transformation), building nucleic acids, proteins and enzyme, facilitating root growth, strengthening stems and stalks and also improving flower formation and seed production. Phosphorus is taken up from the soil solution by plant roots principally as primary orthophosphate ions ($H_2PO_4^-$, pH- 6.5) and to a lesser extent as secondary orthophosphate ions (HPO4²⁻, pH- 7.2). Phosphorus is important for pollination and pollen germination. It also helps in potassium ion translocation, stomatal opening as well it is essential for translocation of sugar. The North-East Region of India has geographical area of 26.3 million hectares and almost 85% of the soils are moderate to strongly acidic. Phosphorus deficiency is the main limiting factor for crop production in acidic soils and therefore, requires the application of phosphatic fertilizers for optimum plant growth and production of food and fibre. An increase in organic phosphorus content decreases the net mineralization but C: P ratio of < 200 enhances organic phosphorus mineralization. The two major forms in which phosphorus is available in the soil are inorganic and organic forms. Indian soils are generally contains 44 to 3580 mg kg⁻¹ of total P (Suresh Kumar, 1999)^[20] and traces to 2160 mg kg⁻¹ of organic P. The organic P can be seen mostly as humus and many other organic materials. The inorganic P occurs in fixed forms or combinations with various forms of AI, Fe, Mg, Ca and other elements mostly unsuitable and unavailable to the plants. Inorganic P (Pi) is the most dominant form of soil P, which constitutes 20-80% of the total P in surface layer (Tomar, 2003). The inorganic phosphorus fraction in soil has been categorized into readily soluble and insoluble categories. 94-99% of the total P is constituted by the insoluble fractions which are not readily available to the plants. This fractions are mostly attached to Fe and Al in acid soils and to Ca in slightly acidic to alkaline soils.

2. Materials and Methods

Surface soil samples (0-15cm) for Investigation were collected from rice cultivated farmer fields of Kakching District of Manipur.

Soil samples collected from different locations of the districts were collected which were dried under shade and further ground in mortar and pestle and passed through 2 mm sieve. These samples were subjected for further analysis for correlation with physico-chemical characteristic. The pH (1:2.5 suspension) and EC (1:2 suspension) were determined as outlined by Jackson, 1973 ^[10], CEC (Borah *et al.*, 1987), organic carbon (Walkley and Black, 1934) ^[23], available nitrogen (Subbiah and Asija, 1956) ^[19], available potassium (Jackson, 1973) ^[10], calcium and magnesium (Chopra and Kanwar, 1976) ^[3] and soil texture (Mechanical analysis, N.B.S.S. and LUP) for each location was calculated. Total phosphorus content of soil sample was determined by Wet digestion method (Jackson, 1973) ^[10].

3. Results and Discussion

Table 1. denotes the physic-chemical properties of the soils from different locations. According to the particle size analysis, mean clay percentage was found to be 56.95%, mean sand percentage was found to be 17.69 and mean silt percentage was found to be 25.37%. The results showed that the mean values of all physico-chemical properties were found to be 4.98 for pH, 0.22 dSm⁻¹ for EC, 14.01 [cmol (p⁺) kg⁻¹] for CEC, 16.76 g kg⁻¹ for organic carbon, 212.66 kg ha⁻¹ for available nitrogen, 114.75 kg ha⁻¹ for available potassium, 2.81 [cmol (p⁺) kg⁻¹] and 1.86 [cmol (p⁺) kg⁻¹] for calcium and magnesium, respectively.

3.1 Inorganic Phosphorus

3.1.1 Saloid phosphorus (Sal-P)

The saloid P content in the soils varied from 4.50 to 15.00 ppm. The average value of saloid P was 7.90 ppm of the studied soils. Saloid P was highly and significantly correlated (Table 3) with E.C (r =0.512*) and C.E.C (r=0.456*) (Malakar *et al.*, 2015) ^[16]. Sal-P was positively and significantly correlated with Ca-P (r=0.520*) and negatively correlated with Org-P (-0.547*).

3.1.2 Aluminium phosphorus (Al-P)

Al-P content in the soils studied was ranged from 25.00 ppm to 85.00 ppm. The mean value of AI-P was 53.10 ppm. Al-P content was and negatively and significantly correlated (Table 3) with pH (r = -0.446*) and positively significantly correlated with CEC (r = 0.699**).

3.1.3 Iron phosphorus (Fe-P)

Fe-P in the soils studied ranged from 40.50 ppm to 112.50 ppm with a mean of 76.26 ppm. Fe-P content was negatively

and significantly correlated (Table 3) with pH (r= -0.447*).

3.1.4 Reductant Soluble phosphorus (Red-P)

Red-P in the studied soils ranged from 51.25 ppm to 110.00 ppm with a mean value of 80.00 ppm. Red-P was positively significantly correlated with CEC (r = 0.555^{*}). A positive and highly significant correlation existed between Red-P and Ca-P (0.628^{**}).

3.1.5 Occluded phosphorus (Occl-P)

Occl-P content in the soil varied from 10.65 ppm to 33.83 ppm average value of Occl-P was 22.02 ppm. Occl-P was negatively and significantly correlated (Table 3) with available potassium (r = -0.505*) and positively and significantly correlated pH (r=0.583**) and CEC (r=0.577**).

3.1.6 Calcium phosphorus (Ca-P)

Amount of Ca-P ranged from 9.23 ppm to 33.20 ppm with a mean value of 21.60 ppm. Ca-P was positively and significantly correlated (Table 4.3) with E.C (r = 0.523* Fig. 11), sand fraction (r=0.500*), clay fraction (r = 0.583**) and CEC (r = 0.605**).

3.2 Organic Phosphorus

The organic-P content in the soil was found in the range from 205.00 ppm to 495.00 ppm with a mean value of 295.38 ppm.

3.3 Total Phosphorus

The total phosphorus content of the soils happened to range from 452.00 to 768.26 ppm. Tot-P was found highly significant with Org-P ($r=0.882^{**}$)

3.4 Tables

Table 1: Physico-chemical properties of the experimental soil

Soil properties	Range	Mean
pH	4.58 - 5.67	4.98
EC (dSm^{-1})	0.14 - 0.31	0.22
CEC [cmol (p^+) kg ⁻¹]	9.00 - 19.00	14.01
Org C (g kg ⁻¹)	10.55 - 26.60	16.76
Available N (kg ha ⁻¹)	75.26 - 305.50	212.66
Available K (kg ha ⁻¹)	49.63 - 160.50	114.75
Ca $[cmol (p^+) kg^{-1}]$	1.25 - 4.40	2.81
Mg [cmol (p^+) k g^{-1}]	0.15 - 4.30	1.86
Sand (%)	6.20 - 50.92	17.69
Silt (%)	8.18 - 42.50	25.37
Clay (%)	19.79 - 83.12	56.95

Table 2: Concentra	ation of different Pfractons in the solis
	nnm

Coil No	ppm									
Son No.	Saloid-P	Al-P	Iron-P	Red-P	Occl-P	Ca-P	Org-P	Total-P		
1	6.50	68.25	97.13	68.50	21.00	19.07	350.00	630.45		
2	1.35	53.83	103.38	59.25	33.83	21.62	495.00	768.26		
3	6.24	41.90	89.45	54.35	23.70	12.67	425.00	653.31		
4	7.35	55.00	87.50	88.80	20.40	26.28	212.50	497.83		
5	6.75	40.10	80.65	97.00	25.10	25.06	325.00	599.66		
6	5.33	32.73	76.00	81.00	23.30	22.70	262.50	503.56		
7	10.00	42.50	50.45	91.00	19.80	26.45	240.00	480.20		
8	12.33	45.50	77.50	75.50	23.85	24.50	245.00	504.18		
9	2.48	34.95	70.80	67.00	16.38	9.23	317.50	518.34		
10	5.50	43.60	84.65	65.85	19.80	16.25	241.00	476.65		
11	10.00	81.50	112.50	98.75	10.65	18.53	312.50	644.43		
12	7.00	49.52	67.52	87.50	12.50	15.84	288.00	527.88		
13	6.53	76.53	82.20	79.00	24.28	22.83	233.50	524.87		

14	12.25	47.50	86.95	51.25	30.05	19.00	205.00	452.00
15	4.50	25.00	40.50	69.00	25.80	21.00	310.00	495.80
16	8.25	61.00	77.25	110.00	19.45	27.50	327.50	630.95
17	15.00	60.00	67.50	95.65	27.05	30.00	240.00	535.20
18	9.80	62.50	45.20	69.30	22.42	14.70	257.50	481.42
19	13.50	85.00	80.00	90.00	18.07	33.20	295.00	614.77
20	7.25	55.00	48.00	101.30	23.00	25.50	325.00	585.05
Mean	7.90	53.10	76.26	80.00	22.02	21.60	295.38	556.24

Table 3: Simple correlation co-efficient between the different forms of phosphorus and soil Properties

Sl. No.		Sal-P	Al-P	Fe-P	Red-P	Occl-P	Ca-P	Org-P	Tot-P
1	pН	-0.065	-0.446*	-0.447*	-0.131	0.583**	0.183	0.201	0.277
2	E.C.	0.512*	0.13	0.211	0.318	-0.091	0.523*	-0.174	-0.134
3	0.C.	-0.068	0.083	0.213	0.365	-0.309	-0.046	0.252	0.224
4	Av. N	-0.119	0.012	0.077	0.168	-0.315	-0.137	0.158	0.174
5	Av. K	-0.022	0.040	-0.301	0.237	-0.505*	-0.139	-0.018	0.831**
6	Sand	-0.04	0.189	-0.189	0.294	-0.16	0.500*	-0.211	0.442
7	Silt	0.104	-0.277	-0.137	-0.331	-0.306	-0.144	-0.295	0.272
8	Clay	-0.032	0.023	0.229	-0.026	-0.063	0.583**	0.344	0.079
9	Ca	0.017	0.025	-0.017	0.018	-0.233	-0.053	0.240	0.251
10	Mg	-0.029	0.053	0.469*	-0.216	0.112	-0.105	0.196	0.257
11	C.E.C.	0.456*	0.699**	0.185	0.555*	0.577**	0.605**	0.164	0.092

**Correlation is significant at the 0.01 level, *Correlation is significant at the 0.05 level

Table 4: Simple correlation co-efficient among the different forms of phosphorus

Sl. No.	Forms	Saloid-P	Al-P	Iron-P	Red-P	Occl-P	Ca-P	Org-P	Total-P
1	Saloid-P	1							
2	Al-P	0.436	1						
3	Iron-P	-0.090	0.394	1					
4	Red-P	0.326	0.337	-0.157	1				
5	Occl-P	-0.079	-0.275	-0.040	-0.436	1			
6	Ca-P	0.520*	0.334	-0.083	0.628**	0.194	1		
7	Org-P	-0.547*	-0.025	0.307	-0.193	0.198	-0.218	1	
8	Total-P	-0.272	0.365	0.540*	0.097	0.098	0.095	0.882**	1

**Correlation is significant at the 0.01 level, *Correlation is significant at the 0.05 level

4. Conclusion

The studied soils were found to have availability of different inorganic forms of phosphorus in the order Red-P> Fe-P> Al-P>Occl-P>Ca-P>Sal-P. Red-P was found to have maximum amount of inorganic phosphorus among the other fractions whereas, Sal-P was to constitute lowest fraction of inorganic phosphorus.

5. Acknowledgements

The first author genuinely wants to put forward his gratitude towards the Dept. of Soil Science and Agricultural Chemisty, College of Agriculture, Central Agricultural University, Imphal for providing opportunities to investigate this research as well as to the farmers who were interactive for the said investigation and helped throughout.

6. Reference

- 1. Banerjee B. Tea production and processing. Oxford and IBH Pub. Co., New Delhi 1993, 188-225.
- 2. Chang SC, Jackson ML. Fractionation of soil phosphorus. Soil Science 1957;84:133-144.
- 3. Chopra SL, Kanwar JS. Analytical Agricultural Chemistry. Kalyani Publisher Ludhiana, New Delhi 1976.
- 4. Dewis J, Freitas F. Physical and chemical methods of soil and water analysis. Oxford and IBH Pub. Co., New Delhi 1984, 51-106.
- 5. Dey SK, Bhattacharyya NG. Studies on soil phosphate. Two Bud 1980;27:21-23.

- 6. Dongale JH. Depthwise distribution of different forms of phosphorus in lateritic soils of coastal region. J Indian Society of Soil Science 1993;41:62-66.
- Havlin JL, Beaton JD, Tisdale SL, Nelson WL. Soil Fertility and fertilizers: An introduction to nutrient management, Sixth Edn. Pearson Edn. Pvt. Ltd., New Delhi 2004, 162.
- Hesse PR. A textbook of soil chemical analysis. John Murray (Publishers) Ltd. 50 Albemarle Street, London, W1X4BD 1971.
- Ibia TO, Udo EJ. Phosphorus forms and fixation of representative soils in Akwa Ibo, State of Nigeria. Geoderma 1993, 95-106.
- Jackson ML. Soil Chemical Analysis. Prentice Hall of India Private Limited, New Delhi 1973.
- 11. Kaistha BP, Sharma PC, Dubey YP. Effect of manure and phosphorus on the inorganic phosphorus fractions in mountain soil of Himalayas. Crop Research 1999;18(2):206-210.
- 12. Khaswa SL, Dubey RK, Tiwari RC, Singh S, Choudhary BS, Singh I. Nitrogen and phosphorous content and uptake, soil nutrient balance and soybean productivity under different levels and sources of phosphorous and plant growth regulators in sub-humid Rajasthan 2014;9(1):1107-1112.
- 13. Kokani JM, Shah KA, Tandel BM, Bhimani GJ. Effect of FYM, phosphorous and sulphur on yield of summer Black gram and post harvest nutrient status of soils. The Bioscan 2015;10(1):379-383.

- 14. Loganathan P, Dayaratne MN, Shanmuganatham. Evaluation of the phosphorus states of some coconut growing soils of Sri Lanka. Journal of Agricultural Science 1982;99:25-33.
- 15. Loganathan P, Sutton PM. Phosphorus fractions and availability in soils formed on different geological deposits in the Niger delta area of Nigeria. Soil Science; 1987;143(1):16-25.
- 16. Malakar H, Gosh D, Chatterjee S, Debnath A. Inorgnaic and organic soil phosphorous fractions in humid tropical tea plantations of West Bengal, India. The Bioscan 2015;10(1):313-318.
- 17. Peterson GW, Corey RB. A modified Chang and Jackson procedure for routine fractionation of inorganic soil phosphates. Soil Science 1966;163:31-36.
- Sarkar D, Haldar A, Mandal D. Forms of phosphorous in relation to soil maturity along a toposequence under hot, dry sub-humid agro-ecological sub-region of West Bengal. Journal of Indian Society of Soil Science 2014;62 (1):29-37.
- 19. Subbiah BV, Asija GL. A rapid procedure for estimation of available N in soils. Current Science 1956;25:259-260.
- Sureshkumar P. Final report of ICAR Adhoc Scheme on variability in Iron and Zinc availability in the Laterite and Lateritic Soils of Central Kerala with Reference to Rice Nutrition. Radiotracer Laboratory, College of Horticultural University, Thrissur 1999.
- 21. Thomas GW, Peaslee DE. Testing soil for phosphorus in soil testing and plant analysis. Walsh and JD Beaton (eds.). Soil Science Society of America, Madison Wisconsin 1973, 115-132.
- 22. Uzu FO, Juo ASR, Fayemi AP. Forms of phosphorus in some important agricultural soils of Nigeria. Soil Science 1975;120:212-218.
- 23. Wakley A, Black IA. An estimation of the Degtjareff method for determining soil organic matter and a proposed modification of the chromic acid titration method. Soil Science 1934;37:29-38.