



ISSN (E): 2277-7695  
ISSN (P): 2349-8242  
NAAS Rating: 5.23  
TPI 2023; 12(2): 1158-1162  
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www.thepharmajournal.com  
Received: 16-12-2022  
Accepted: 27-01-2023

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## Synthesis, characterization and evaluation of bactericidal activity of silver nano particle

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### Abstract

The current study was undertaken to explore the effect of silver nano particle as a topical bactericidal agent. Silver nano particle (AgNP) was synthesized by chemical reduction method and characterized by Transmission Electron Microscope (TEM). The bactericidal activity of AgNP was determined using the disc, gauze and well diffusion method against common clinical pathogenic bacteria. The highest bactericidal effect was observed at the concentration of 40  $\mu$ L as a maximum diameter of zone of inhibition (mm) and the observed results showed that the AgNPs have potent bactericidal activity against both gram positive and gram negative bacteria. The present study indicated that AgNP can effectively be used as an alternate to antibiotic topical agent, when antibiotic resistance is suspected.

**Keywords:** Silver nano particle, synthesis, characterization, pathogenic bacteria, bactericidal activity

### Introduction

Among the most commonly used topical antimicrobial agent, silver based topical application is employed due to its broad spectrum activity, efficacy and lower costs with low toxicity. Increased therapeutic use of antibiotics resulted in the emergence of antibiotic resistant bacteria especially multidrug resistant (MDR) bacteria, a major continuous threat to treatment. Silver, and its derivatives, are known strong effective agent against bacteria, fungi and viruses (Lok, *et al.*, 2006) [12]. To curtail the development of antibiotic resistance, various nano metal preparations like silver, gold, copper, zinc have been incorporated in many pharmaceutical preparations for their catalytic, optical, electronic, magnetic and anti-microbial properties (Cao, 2004 and Cho, 2005) [3, 4]. Several previous studies have shown that silver nanoparticles (Ag-NPs) are effectively incorporated into wound and burn dressings (Jia, *et al.*, 2007) [10]. Gemmell *et al.*, 2006 reported that Ag-NPs were effective against *Pseudomonas aeruginosa*, ampicillin resistant *Escherichia coli*, erythromycin-resistant *Streptococcus pyogenes*, methicillin-resistant *Staphylococcus aureus* (MRSA) and vancomycin-resistant *Staphylococcus aureus* (VRSA). Though various methods are employed for the synthesis of AgNPs (Gudikandula and Maringanti, 2016) [8], chemical method proved to be nontoxic and cost-effective. In the current study, a chemical reduction method is used for the synthesis of AgNPs and is evaluated for the bactericidal activity after morphological characterization.

### Materials and Methods

The chemical reagents - Silver nitrate and tri sodium citrate were purchased from Sigma-Aldrich and all the reagents required for the study were prepared in double distilled water.

#### 1. Chemical Synthesis of Silver Nanoparticles

The silver nano particle was prepared by chemical reduction method as per Fang *et al.*, (2005) [5]. A stock aqueous solution of 1mM AgNO<sub>3</sub> and 1% tri sodium citrate solution were prepared. A working solution of 500 ml of AgNO<sub>3</sub> was heated to boiling. To this solution, 10 ml of trisodium citrate was added drop by drop with vigorous magnetic stirring continuously for 30min. and heated until noticeable colour change. This indicates the conversion of AgNO<sub>3</sub> into AgNP. Green-gray silver nano particle were obtained immediately after cooling the solution.

#### 2. Characterization of Silver Nanoparticles

The identity and size of obtained Ag-NPs was confirmed by ultraviolet-visible spectrophotometer (UV-Vis Spectrophotometer, PerkinElmer LAMBDA, Waltham, USA) and characterized by transmission electron microscopy (TEM).

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### 3. Bactericidal activities of Silver nanoparticles

The bacterial strains *viz.*, gram-positive (*Staphylococcus aureus*) and gram-negative (*Escherichia coli*) bacteria were obtained from multidrug-resistant (MDR) clinical pathogenic isolates and passed three times in nutrient broth medium to optimize their physiological activities. The bactericidal activity studies were performed by three methods as follows in triplicate.

#### 1. Paper disc diffusion method

Using disc diffusion method, the bactericidal activity of AgNP was determined. The bacterial strains *viz.*, gram-positive (*Staphylococcus aureus*) and gram-negative (*Escherichia coli*) were tested against the synthesized AgNP. A range of concentrations of AgNP were prepared (10, 20, 30, 40 and 50  $\mu$ L) and paper discs (5 mm) were soaked in each concentration. Using a glass rod, the bacterial strain (100 mL) was inoculated by smearing on the medium and dried for 5 min. The synthesized AgNP loaded discs were placed on the surface of the medium and allowed to diffuse for 5 min. The petridish were incubated at 37° C for 24 h and were then examined for the presence of zones of inhibition (mm). A disc soaked in sterilized water (20 $\mu$ L) was used as a negative control and silver nitrate (AgNO<sub>3</sub>) disc (20  $\mu$ L and 40  $\mu$ L /disc) was used as a positive control. After deducting the diameter of the positive control zones from the AgNP zone and the obtained zone of inhibition diameter, expressed in millimeter unit (mm) was tabulated.

#### 2. Agar well-diffusion method

Briefly, 20 mL of nutrient agar medium was poured into sterilized Petri dishes. To prepare the bacterial colonies (Gram-positive and Gram-negative bacteria) of 1 x 10<sup>5</sup> CFU/mL one day old bacterial cultures was used. Agar wells (8 mm diameter) were prepared with the help of a sterilized template. The wells were injected with a range of 5, 10, 20, 30, 40 and 50  $\mu$ L concentrations of synthesized AgNP solution. The plates were further incubated at 37° C for 24 h and examined for the presence of zones of inhibition and measured (mm).

#### 3. Gauze diffusion method

After confirming the antimicrobial potential of AgNP using disc and well diffusion methods, the effective dose was incorporated into a gauze cloth. As detailed above, instead of paper disc, sterile surgical gauze of 5 mm X 5mm size was soaked in 40  $\mu$ L synthesized AgNP solution, 40  $\mu$ L of 1 mM AgNO<sub>3</sub> as a positive control, 40  $\mu$ L of double distilled water used in the experiment as a negative control along with a sterile surgical gauze as a blank. Aliquots of bacterial culture (100 mL) were spread on petri dishes containing agar-solidified Luria broth (LB) medium. The prepared and loaded gauze as specified above, were placed and incubated at 37° C for 24 h. The presence of zones of inhibition (mm) was recorded.

## Results and Discussion

### 1. Chemical synthesis of silver nanoparticles

The silver nanoparticles are formed by the chemical reduction of AgNO<sub>3</sub>, with a powerful reducing and stabilizing agent – Tri sodium citrate. The formed AgNP colloids are affected by initial concentration of AgNO<sub>3</sub> (molar ratio) and stabilizer concentration. Among the various methods available for the

synthesis of silver nanoparticles, the best and most easy economical high yielding method without aggregation is chemical reduction method (Irvani *et al.*, 2014)<sup>[9]</sup>. Green-gray silver nano particle were obtained immediately after cooling the solution (Fig. 1).

### 2. Characterization of Silver Nanoparticles

The synthesized Ag-NP colloids were analyzed for absorbance at 300–700 nm in a UV-visible (UV-Vis) spectrophotometer. The identity of synthesized Ag-NP colloids by UV– visible spectrophotometer absorption spectra had a peak value of 420 nm (Fig. 2).

Earlier reports of optical measurements of Ag-NP confirmed by UV-Vis spectrophotometer analysis showed an absorbance peak at 420 nm which was specific for silver nanoparticles (Gudikandula and Maringanti, 2016)<sup>[8]</sup> whereas Sayed *et al.* (2015)<sup>[17]</sup> reported the biosynthesis of Ag nanoparticles using *Aspergillus sp.* in the range of 525 nm.

Stable and mono dispersed AgNP synthesized chemically were recovered and characterized by transmission electron microscopy (Fig. 3). The identity and size of synthesized AgNP were spherical and are in a range of 43.82 to 73.24 nm. The resultant TEM images are in agreement with the reports of Panacek *et al.* (2006)<sup>[14]</sup>, Anil Kumar *et al.* (2007)<sup>[2]</sup> and Galdiero *et al.* (2011)<sup>[6]</sup> who showed spherical shaped chemically synthesized silver nanoparticles in a range of 10 to 30 nm while Abalkhila *et al.*, 2017<sup>[1]</sup> recorded 10-60 nm spherical sized AgNP using extracts from *A. vera*, *P. oleracea* and *C. dactylon*. Similarly, Irvani *et al.*, 2014<sup>[9]</sup> reported chemically synthesized AgNP with tri sodium citrate as reducing cum stabilizing agent yield a nano particle size of 30-60 nm.

The synthesis of Ag-NP and characterization of Ag-NP by TEM and UV–visible spectrophotometer revealed spherical nanoparticles exhibiting a range of 40-75 nm sizes at 405-420 nm.

### 3. Bactericidal activity studies of Silver nanoparticles

Results from antibacterial activity assessment of AgNP as diameter of inhibition zones (mm) are presented in Table 1 A, B, C and depicted in Fig.4 A, B, C.

Evaluation of the bactericidal activities of Ag-NP revealed that the inhibition zones (mm) around Ag-NPs-loaded discs found to have potent bactericidal activities against both Gram positive and Gram negative bacteria. The highest effect was observed at a concentration of 40  $\mu$ L in disc and gauze diffusion studies. In well diffusion method both 40  $\mu$ L and 50  $\mu$ L concentration range gave almost same diameter of zone of inhibition. The diameter of the inhibition zone was smaller at a concentration of 5.0  $\mu$ L. These results are similar to those reported by other investigators (Abalkhila *et al.*, 2017, Morones *et al.*, 2005 and Kim *et al.*, 2007)<sup>[1, 13, 11]</sup>.

Previous studies of Sondi *et al.* (2004)<sup>[18]</sup>, Sarkar *et al.* (2007)<sup>[16]</sup> and Sayed *et al.* (2015)<sup>[17]</sup> indicated that the synthesized nanoparticles using *aloe vera* and *E. coli* have antimicrobial activity against a wide range of microorganisms.

### Summary

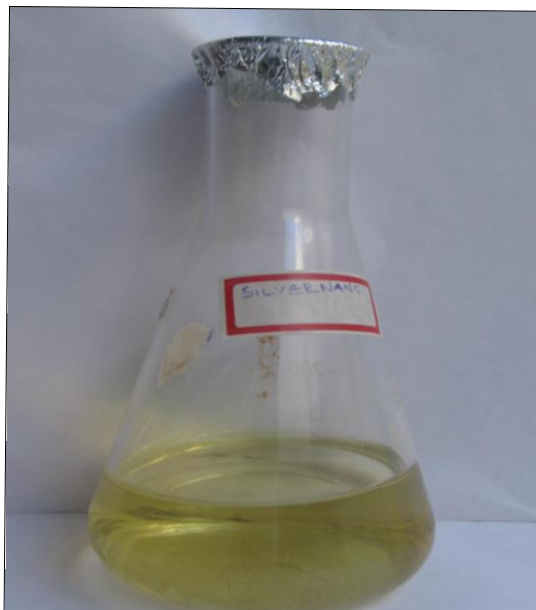
The chemically synthesized of Ag-NP was found to have potent bactericidal activities against both Gram positive and Gram negative bacteria. The results clearly indicate that silver nanoparticles could provide a safer alternative to conventional antimicrobial agents in the form of a topical pharmaceutical

formulation.

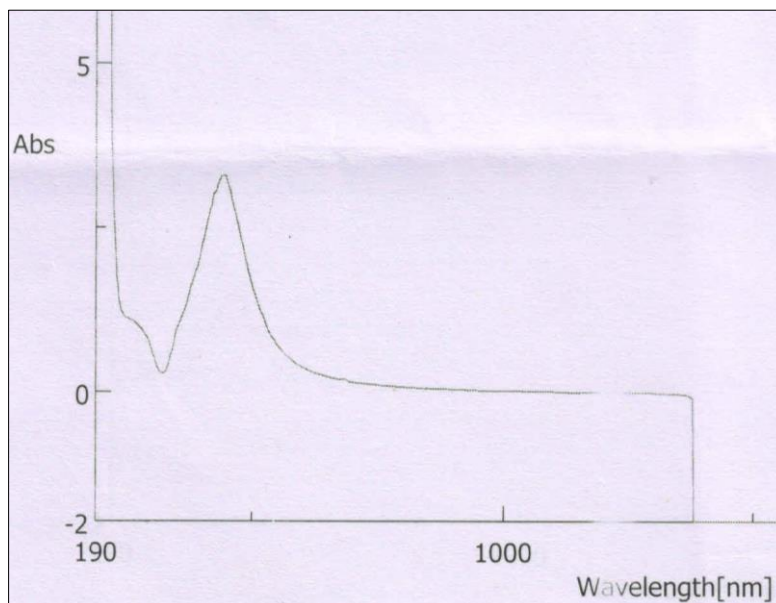
**Conclusion**

The present study provides an evidence for an easy and cheap

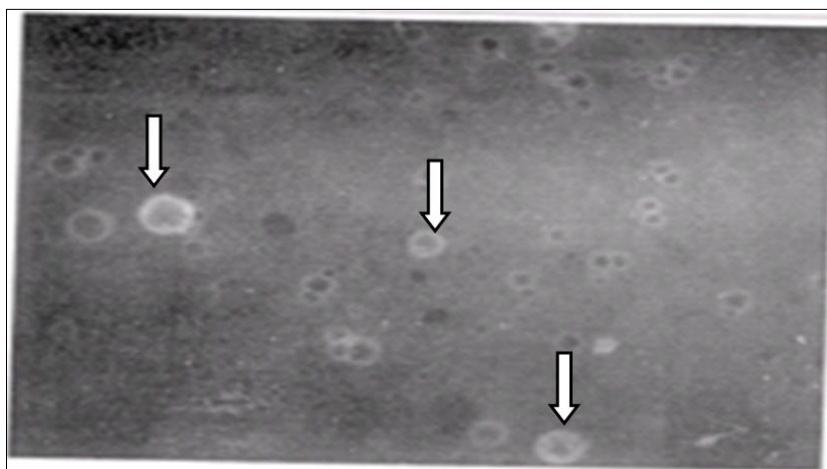
method for synthesizing potent bactericidal Ag-NPs and demonstrates their effectiveness against common pathogenic bacteria as an alternative approach to antibiotics, when antibiotic resistance is suspected.



**Fig 1:** Chemical synthesis of AgNPs



**Fig 2:** UV-Vis Spectrophotometer analysis of chemically synthesized AgNPs



**Fig 3:** TEM images of chemically synthesized AgNPs.

**Table 1:** Antibacterial activity of chemically synthesized Ag-NPs  
A. Disc diffusion method

Bacterial strain	Zone of inhibition (mm diameter)						
	Concentration						Sterile water (Negative control) 20µl
	AgNP(µl)					Silver nitrate (AgNO <sub>3</sub> )	
	10	20	30	40	50	20µl	40µl
Gram-positive ( <i>Staphylococcus aureus</i> )	13.1	12.9	15.4	18.2	17.7	14.2 (C <sub>1</sub> )	17.4 (D)
Gram-negative ( <i>Escherichia coli</i> )	3.1	3.9	4.7	5.8	5.2	2.7 (D)	4.4 (C <sub>1</sub> )


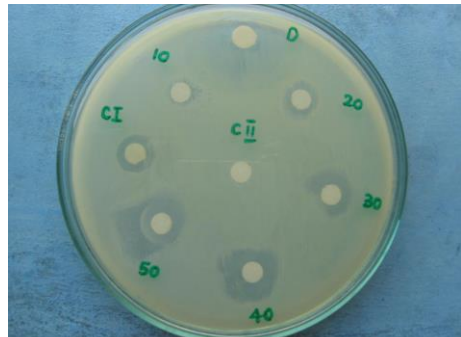
B. Well diffusion method

Bacterial strain	Zone of inhibition (mm diameter)					
	Concentration of AgNP (µl)					
	5	10	20	30	40	50
Gram-positive ( <i>Staphylococcus aureus</i> )	0	12.4	12.4	12.6	13.6	13.6
Gram-negative ( <i>Escherichia coli</i> )	0	2.8	3.3	3.5	4.3	4.2

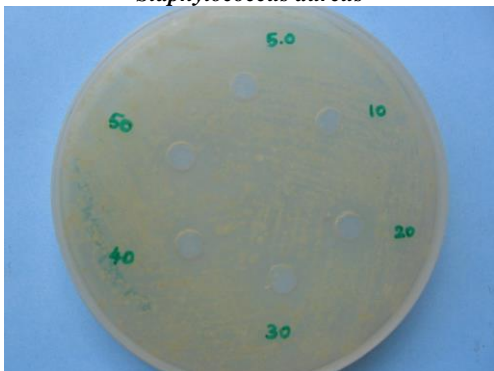
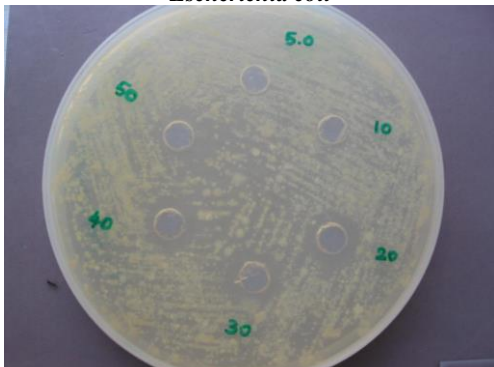
C. Gauze diffusion method

Zone of inhibition (mm diameter)				
Bacterial strain	Blank (sterile gauze) (CIII)	Sterile water (Negative control) 40µL (CI)	Silver nitrate (AgNO <sub>3</sub> )- 40µL (Positive control) (CII)	Silver Nanoparticle (AgNP) (40µL) (SII)
Gram-positive ( <i>Staphylococcus aureus</i> )	0	0	3.8	4.3
Gram-negative ( <i>Escherichia coli</i> )	0	0	0.7	2.7

A. Disc diffusion method

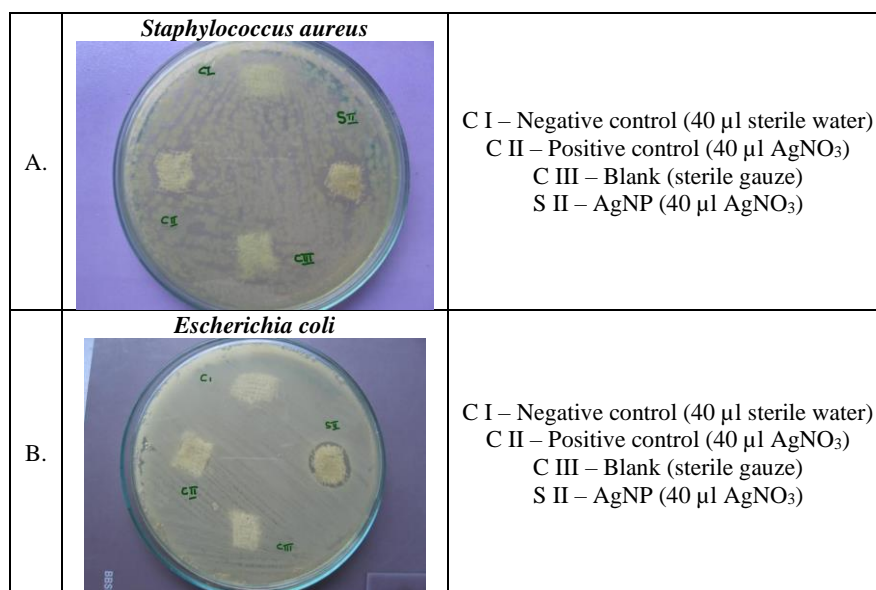
A.	<p><i>Staphylococcus aureus</i></p> 	<p>C II – Negative control (20 µl sterile water)                      C I – Positive control (20 µl AgNO<sub>3</sub>)                      D – Positive control (40 µl AgNO<sub>3</sub>)</p>
B.	<p><i>Escherichia coli</i></p> 	<p>C II – Negative control (20 µl sterile water)                      C I – Positive control (40 µl AgNO<sub>3</sub>)                      D – Positive control (20 µl AgNO<sub>3</sub>)</p>

B. Well diffusion method

A.	<p><i>Staphylococcus aureus</i></p> 	<p>Concentration of AgNP (µl)                      Range - 5,10,20,30,40,50</p>
B.	<p><i>Escherichia coli</i></p> 	<p>Concentration of AgNP (µl)                      Range - 5,10,20,30,40,50</p>



## c. Gauze diffusion method



**Fig 4:** Antibacterial activity of chemically synthesized Ag-NPs

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